

## Fernene Triterpenoids from the Lichen *Pyxine berteriana*

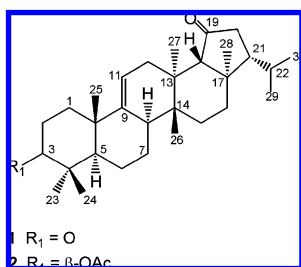
Marta S. Maier,<sup>\*,†</sup> María L. Rosso,<sup>†,‡</sup> Alejandra T. Fazio,<sup>†,‡</sup> Mónica T. Adler,<sup>†</sup> and María D. Bertoni<sup>‡</sup>

UMYFOR (CONICET–UBA) and Departamento de Química Orgánica, Facultad de Ciencias Exactas y Naturales, and PROPLAME-PRHIDEB-Departamento de Biodiversidad y Biología Experimental, Facultad de Ciencias Exactas y Naturales, Pabellón 2, Ciudad Universitaria, 1428 Buenos Aires, Argentina

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Two new fernene triterpenoids, fern-9(11)-en-3,19-dione (**1**) and 3 $\beta$ -acetoxyfern-9(11)-en-19-one (**2**), together with the known 3 $\beta$ -acetoxyfern-9(11)-en-19 $\beta$ -ol (**3**) and lichexanthone (**4**), have been isolated from the acetone extract of the lichen *Pyxine berteriana*. The structures of the new compounds were established on the basis of IR, extensive 1D and 2D NMR, and MS analyses. Although several fern-9(11)-enes have been isolated from lichens, compounds **1** and **2** are the first examples of naturally occurring fernene triterpenoids with a carbonyl function at C-19.

Lichens are symbiotic associations composed of at least a fungal partner, the mycobiont, and a photosynthetic partner, the photobiont.<sup>1</sup> These associations frequently produce characteristic secondary metabolites that are of fungal origin. Most are unique to lichens, and only a small number occur in non-lichenized fungi or higher plants.<sup>2</sup> Many of these lichen secondary compounds exhibit antibiotic, antitumor, antimutagenic, allergenic, antifungal, antiviral, enzyme inhibitory, and plant growth inhibitory properties.<sup>2,3</sup> Triterpenoids are widely distributed in lichens, being commonly present in genera such as *Nephroma* and *Pseudocyphellaria* as well as in different genera of the Physciaceae (e.g., *Dirinaria*, *Physcia*, and *Pyxine*) and the Parmeliaceae (e.g., *Parmelia* and *Evernia*).<sup>3,4</sup> A previous report on the secondary metabolites of *Pyxine berteriana* (Physciaceae) from Brazil indicated that it contained atranorin, lichexanthone, methyl pyxinate, and pyxinol, according to TLC analysis.<sup>4</sup> In the course of the search for new metabolites from the lichen *P. berteriana* (Fée) Imshaug we have isolated two new fernene triterpenoids, fern-9(11)-en-3,19-dione (**1**) and 3 $\beta$ -acetoxyfern-9(11)-en-19-one (**2**), together with the known 3 $\beta$ -acetoxyfern-9(11)-en-19 $\beta$ -ol (**3**) and lichexanthone (**4**), which is a chemical marker of a group of species in the genus *Pyxine*.<sup>5,6</sup> The structure elucidation of compounds **1** and **2** is described herein.



Compound **1** was obtained as a white, amorphous powder and showed a molecular ion at  $m/z$  438.3485 in the HREIMS, indicative of a molecular formula of  $C_{30}H_{46}O_2$ . Its IR absorption bands at 1731 and 1708  $cm^{-1}$  suggested the presence of two ketone groups contained in a cyclopentanone<sup>7</sup> and a cyclohexanone ring,<sup>8</sup> respectively. The assignment of  $^1H$  and  $^{13}C$  NMR spectroscopic data of **1** (Table 1) was based on DEPT, HSQC, HMBC, NOESY, and  $^1H$ – $^1H$  COSY spectra. A DEPT NMR experiment permitted differentiation of the 30  $^{13}C$  NMR resonances into eight methyl,

eight  $sp^3$  methylene, five  $sp^3$  methine, and five  $sp^3$  quaternary carbons, in addition to two carbonyls ( $\delta_C$  215.4 and 216.6) and a trisubstituted vinyl group resonating at  $\delta_H$  5.39 and at  $\delta_C$  117.1 and 148.9, characteristic of a  $\Delta^{9,11}$  double bond.<sup>9</sup> Characteristic resonances in the  $^1H$  and  $^{13}C$  NMR spectra (Table 1) for six tertiary methyls [ $\delta_{H/C}$  0.77/15.6 (C-26), 0.91/15.8 (C-28), 0.97/17.0 (C-27), 1.04/24.3 (C-23), 1.12/21.7 (C-24), 1.30/24.2 (C-25)] and two secondary methyls [ $\delta_{H/C}$  0.88/22.8 (C-30), 0.99/22.2 (C-29)] indicated a fern-9(11)-ene-type pentacyclic triterpenoid skeleton.<sup>9</sup> In accordance with the COSY spectrum, the signal at  $\delta_H$  2.76 (H-2 $\beta$ ) showed cross-peaks with the signals at  $\delta_H$  2.20 (H-1 $\beta$ ), 2.23 (H-2 $\alpha$ ), and 1.64 (H-1 $\alpha$ ). On the basis of the HMBC and HSQC spectra, the signals at  $\delta_H$  2.76 and 2.23 ( $\delta_C$  35.1, H-2) showed cross-peaks with the signals at  $\delta_C$  40.4 (C-1), 216.6 (C-3), and 37.6 (C-10), establishing that C-3 corresponded to the carbonyl group at  $\delta_C$  216.6. Further correlations in the HMBC spectrum of the singlet at  $\delta_H$  1.30 ( $\delta_C$  24.2) with the signals at  $\delta_C$  37.6 (C-10), 40.4 (C-1), 46.4 (C-5), and 148.9 (C-9) allowed us to assign this methyl resonance to C-25. The NOESY correlations between H-5/CH<sub>3</sub>-23 and CH<sub>3</sub>-25/CH<sub>3</sub>-24 in conjunction with HSQC data permitted assignment of the  $^1H$  and  $^{13}C$  resonances of CH<sub>3</sub>-23 ( $\delta_H$  1.04,  $\delta_C$  24.3) and CH<sub>3</sub>-24 ( $\delta_H$  1.12,  $\delta_C$  21.7). HMBC correlations of these methyl protons with the signal at  $\delta_C$  216.6 confirmed the assignment of this carbonyl group to C-3. The broad doublet at  $\delta_H$  2.09 ( $\delta_C$  38.8) was assigned to H-8 on the basis of the cross-peaks with the signals at  $\delta_H$  1.41 and 1.70 (H-7) in the COSY spectrum, while correlations of the signal at  $\delta_H$  2.09 with H-5 (1.72) and CH<sub>3</sub>-27 ( $\delta_H$  0.97) in the NOESY spectrum confirmed its  $\alpha$ -orientation. We assigned the chemical shift of C-18 at  $\delta_C$  64.6 on the basis of the HMBC correlations of CH<sub>3</sub>-27 ( $\delta_H$  0.97) and CH<sub>3</sub>-28 ( $\delta_H$  0.91) to C-18. The NOESY cross-peaks between H-18 ( $\delta_H$  2.18) and CH<sub>3</sub>-26 ( $\delta_H$  0.77) showed that H-18 had a  $\beta$ -orientation and indicated *trans*-fusion of the D/E ring. On the other hand, H-18 correlated in the HMBC spectrum with the signals at  $\delta_C$  36.3 (C-13), 43.0 (C-17), and the carbonyl group at  $\delta_C$  215.4. This observation together with the presence of a singlet for H-18 in the  $^1H$  NMR spectrum suggested that C-19 corresponded to the carbonyl group at  $\delta_C$  215.4. Further HMBC correlations of H-20 $\beta$  ( $\delta_H$  2.38) to C-17, C-19, and C-21 confirmed the presence of a carbonyl group at C-19. On the basis of the NOESY spectrum, the signal at  $\delta_H$  1.45 ( $\delta_C$  55.2, H-21) showed cross-peaks with H-18 $\beta$ , H-20 $\beta$ , and CH<sub>3</sub>-26, confirming the  $\beta$ -orientation of H-21, and hence the  $\alpha$ -orientation of the isopropyl chain. Thus, this natural product corresponds to fern-9(11)-en-3,19-dione.

Compound **2** was obtained as a white, amorphous powder. Its HREIMS showed a molecular ion at  $m/z$  482.3782, indicative of a molecular formula of  $C_{32}H_{50}O_3$  and eight unsaturation degrees. Three of these were due to the presence of two carbonyl groups

\* Corresponding author. Tel and /Fax: +54 11 4576-3385. E-mail: maier@qo.fcen.uba.ar.

<sup>†</sup> UMYFOR (CONICET–UBA) and Departamento de Química Orgánica.

<sup>‡</sup> PROPLAME-PRHIDEB-Departamento de Biodiversidad y Biología Experimental.

**Table 1.** NMR Spectroscopic Data (500 MHz, CDCl<sub>3</sub>) of Compounds **1** and **2**<sup>a</sup>

position	<b>1</b>		<b>2</b>	
	$\delta_C$ , mult.	$\delta_H$ (J in Hz)	$\delta_C$ , mult.	$\delta_H$ (J in Hz)
1	40.4, CH <sub>2</sub>	1.64, m (H-1 $\alpha$ ) 2.20, m (H-1 $\beta$ )	39.0, CH <sub>2</sub>	1.40, m (H-1 $\alpha$ ) 1.93, dt (13.6, 3.4) (H-1 $\beta$ )
2	35.1, CH <sub>2</sub>	2.23, m (H-2 $\alpha$ ) 2.76, td (14.6, 5.6) (H-2 $\beta$ )	24.6, CH <sub>2</sub>	1.66, m
3	216.6, qC		80.9, CH	4.48, dd (9.0, 6.7)
4	48.0, qC		38.1, qC	
5	46.4, CH	1.72, m	44.5, CH	1.38, m
6	19.2, CH <sub>2</sub>	1.40 (H-6 $\alpha$ ), 1.75 (H-6 $\beta$ ), m	18.8, CH <sub>2</sub>	1.61, 1.76, m
7	17.8, CH <sub>2</sub>	1.41, 1.70, m	17.8, CH <sub>2</sub>	1.27, 1.63, m
8	38.8, CH	2.09, bd (13.6)	39.0, CH	2.02, bd
9	148.9, qC		150.1, qC	
10	37.6, qC		37.6, qC	
11	117.1, CH	5.39, m	116.4, CH	5.32, m
12	35.0, CH <sub>2</sub>	2.51, ddd (17.9, 5.6, 1.5) (H-12 $\alpha$ ), 1.68, m (H-12 $\beta$ )	35.1, CH <sub>2</sub>	2.49, ddd (17.8, 5.4, 1.8) (H-12 $\alpha$ ), 1.66, m (H-12 $\beta$ )
13	36.3, qC		36.3, qC	
14	37.3, qC		37.3, qC	
15	28.9, CH <sub>2</sub>	1.40, m	28.9, CH <sub>2</sub>	1.36, m
16	35.8, CH <sub>2</sub>	1.65, m (H-16 $\beta$ ), 1.82, dt (13.2, 3.4) (H-16 $\alpha$ )	35.8, CH <sub>2</sub>	1.64, m (H-16 $\beta$ ), 1.83, dt (13.2, 3.3) (H-16 $\alpha$ )
17	43.0, qC		43.1, qC	
18	64.6, CH	2.18, s	64.7, CH	2.17, s
19	215.4, qC		215.5, qC	
20	42.4, CH <sub>2</sub>	1.77, dd (18.9, 10.3) (H-20 $\alpha$ ), 2.38, dd (18.9, 8.2) (H-20 $\beta$ )	42.4, CH <sub>2</sub>	1.77, dd (18.9, 10.0) (H-20 $\alpha$ ), 2.38, dd (18.9, 8.0) (H-20 $\beta$ )
21	55.2, CH	1.45, m	55.2, CH	1.44, m
22	30.2, CH	1.64, m	30.2, CH	1.63, m
23	24.3, CH <sub>3</sub>	1.04, s	27.4, CH <sub>3</sub>	0.84, s
24	21.7, CH <sub>3</sub>	1.12, s	16.1, CH <sub>3</sub>	0.94, s
25	24.2, CH <sub>3</sub>	1.30, s	25.2, CH <sub>3</sub>	1.08, s
26	15.6, CH <sub>3</sub>	0.77, s	15.6, CH <sub>3</sub>	0.73, s
27	17.0, CH <sub>3</sub>	0.97, s	16.9, CH <sub>3</sub>	0.98, s
28	15.8, CH <sub>3</sub>	0.91, s	15.8, CH <sub>3</sub>	0.91, s
29	22.2, CH <sub>3</sub>	0.99, d (6.6)	22.3, CH <sub>3</sub>	0.98, d (6.5)
30	22.8, CH <sub>3</sub>	0.88, d (6.6)	22.8, CH <sub>3</sub>	0.88, d (6.5)
CH <sub>3</sub> COO			170.9, qC	
CH <sub>3</sub> COO			21.3, CH <sub>3</sub>	2.05, s

<sup>a</sup> Assigned by a combination of <sup>1</sup>H–<sup>1</sup>H COSY, NOESY, HSQC, and HMBC experiments.

(one band at 1732 cm<sup>-1</sup> in the IR consistent with both a cyclopentanone ring and an ester group at  $\delta_C$  170.9 and 215.5 in the <sup>13</sup>C NMR spectrum) and one trisubstituted double bond [ $\delta_H$  5.32 and  $\delta_C$  116.4 and 150.1]. The assignment of <sup>1</sup>H and <sup>13</sup>C NMR spectroscopic data of **2** (Table 1) was based on DEPT, HSQC, HMBC, NOESY, and <sup>1</sup>H–<sup>1</sup>H COSY spectra. The <sup>13</sup>C NMR spectrum showed 32 resonances, of which 30 were attributed to a fern-9(11)-ene-type pentacyclic triterpene skeleton and two to an acetyl group ( $\delta_H$  2.05;  $\delta_C$  21.3 and 170.9).<sup>9</sup> A DEPT NMR experiment showed the presence of nine methyl, eight sp<sup>3</sup> methylene, six sp<sup>3</sup> methine (one oxygenated), and five sp<sup>3</sup> quaternary carbons, in addition to the carbonyl and olefinic carbons. Analysis of the <sup>13</sup>C NMR spectroscopic data of **2** (Table 1) revealed structural similarity to those of 3 $\beta$ -acetoxyfern-9(11)-en-19 $\beta$ -ol (**3**),<sup>9</sup> except for the presence of a ketone functionality ( $\delta$  215.5 ppm) and the absence of the hydroxy group at C-19 ( $\delta_C$  71.2 ppm,  $\delta_H$  4.23 ppm). This observation, along with the HMBC correlations from H-18 ( $\delta_H$  2.17) to C-13, C-17, C-28, and the carbonyl signal at  $\delta_C$  215.5, indicated the presence of a C-19 ketone functionality. Further HMBC correlations from H-20 to C-17, C-21, C-22, and the signal at  $\delta_C$  215.5 together with comparison of <sup>1</sup>H and <sup>13</sup>C NMR data of **2** with those of rings C, D, and E of compound **1** confirmed that both compounds shared the same ketone functionality at C-19 and differed in the substituent at C-3. The NMR data of compound **2** showed close resemblance with those of rings A and B and CH<sub>3</sub>-23, CH<sub>3</sub>-24, and CH<sub>3</sub>-25 of 3 $\beta$ -acetoxyfern-9(11)-en-19 $\beta$ -ol (**3**).<sup>9</sup> On the basis of the HMBC and HSQC spectra, the signals at  $\delta_H$  1.40 and 1.93 ( $\delta_C$  39.0, H-1) showed cross-peaks with the signals at  $\delta_C$  24.6 (C-2), 25.2 (CH<sub>3</sub>-25), 37.6 (C-10), and 44.5 (C-5), whereas the signal at  $\delta_H$  4.48 ( $\delta_C$  80.9, H-3) correlated with the signals at  $\delta_C$  16.1 (CH<sub>3</sub>-24), 24.6 (C-2), 27.4 (CH<sub>3</sub>-23), and 38.1 (C-4). These data together with the HMBC correlations of the methyl singlet at  $\delta_H$  2.05 ( $\delta_C$  21.3, CH<sub>3</sub>COO) with  $\delta_C$  80.9 (C-3) and 170.9 (CH<sub>3</sub>COO) established the position of the acetoxy group

at C-3. NOESY correlations between H-3 and  $\delta_H$  0.84 (CH<sub>3</sub>-23), 1.38 (H-5 $\alpha$ ), and 1.40 (H-1 $\alpha$ ) confirmed the  $\beta$ -orientation of the acetoxy group. The NOESY cross-peaks between H-5/CH<sub>3</sub>-23 and CH<sub>3</sub>-25/CH<sub>3</sub>-24 in conjunction with HSQC data allowed assignment of the <sup>1</sup>H and <sup>13</sup>C NMR resonances of CH<sub>3</sub>-23 ( $\delta_H$  0.84,  $\delta_C$  27.4), CH<sub>3</sub>-24 ( $\delta_H$  0.94,  $\delta_C$  16.1), and CH<sub>3</sub>-25 ( $\delta_H$  1.08,  $\delta_C$  25.2). As a consequence, compound **2** is identified as 3 $\beta$ -acetoxyfern-9(11)-en-19-one.

Compounds **1** and **2** are the first naturally occurring examples of fernene-type triterpenoids containing an unprecedented ketone function at C-19. Previously, two semisynthetic derivatives with a carbonyl group at C-19, fern-9(11)-en-19-one and fern-7-en-19-one, were obtained by CrO<sub>3</sub> oxidation of two fernene triterpenoids isolated from the rhizomes of *Davallia solida*.<sup>7</sup> Compound **1** is the second example of a natural fernene-type triterpenoid with two ketone groups. Previously, fern-9(11)-en-3,12-dione has been isolated from the lichen *Xanthoria resendei*.<sup>8</sup>

Compound **3** was isolated as a white, amorphous powder and identified as 3 $\beta$ -acetoxyfern-9(11)-en-19 $\beta$ -ol by comparison of the NMR and EIMS data with those reported previously.<sup>9</sup> Lichexanthone (**4**) was isolated as a yellow, amorphous powder and identified by EIMS and <sup>1</sup>H NMR data and comparison with published data.<sup>3</sup>

## Experimental Section

**General Experimental Procedures.** Optical rotations were measured on a Perkin-Elmer 343 polarimeter. IR spectra were recorded on a Nicolet Magna-550 FT-IR spectrometer. <sup>1</sup>H and <sup>13</sup>C NMR spectra were recorded in CDCl<sub>3</sub> on a Bruker AM 500 spectrometer. EIMS data were recorded on a Shimadzu QP-5000 mass spectrometer. HREIMS were obtained on a VG ZAB T4 mass spectrometer. Analytical HPLC was carried out on a Gilson 506C HPLC chromatograph using a reversed-phase analytical column (Phenomenex Hypersil; 5  $\mu$ m pore size, 250  $\times$  4.6 mm). The samples were eluted with a two-solvent system at a rate of 1 mL min<sup>-1</sup>. Solvent A was 1% aqueous

orthophosphoric acid–MeOH (7:3), and solvent B was MeOH. The gradient started with 0% B and increased to 58% B within 15 min, then to 100% B within 15 min, followed by 100% B for 10 min. Compounds were detected using a 170 photodiode array detector set at 245 nm, operated in series with Unipoint System software, recording the absorption spectrum in the range 200–400 nm. TLC was performed on precoated Si gel F254 (cyclohexane–EtOAc (9:1)) and detected by spraying with H<sub>2</sub>SO<sub>4</sub> (5% EtOH).

**Lichen Material.** Thalli of *P. berteriana* were collected on *M. azedarach* by one of the authors (M.T.A.) at Glew, Buenos Aires Province, Argentina, on October 7, 2000. A voucher specimen (39315) was identified by M.T.A. and preserved at the Herbarium of the Faculty of Exact and Natural Sciences (BAFC), Buenos Aires, Argentina.

**Extraction and Isolation.** The lichen (1.05 g wet weight) was cleaned, cut into small pieces, and extracted in acetone (100 mL) at room temperature. The acetone extract was evaporated under reduced pressure to give a residue (103 mg), which was subjected to silica gel column chromatography using cyclohexane and cyclohexane–EtOAc mixtures (99:1 and 98:2) as eluents to give the pure triterpenoids **1** (8.0 mg), **2** (3.6 mg), and **3** (2.9 mg) and lichexanthone (**4**) (14.6 mg). Compounds **1** and **2** showed one peak in their HPLC chromatograms at *t*<sub>R</sub> 41.4 and 37.5 min, respectively.

**Fern-9(11)-en-3,19-dione (1):** white, amorphous powder; [ $\alpha$ ]<sub>D</sub><sup>20</sup> –9.3 (c 0.33, CHCl<sub>3</sub>); IR (KBr)  $\nu_{\max}$  2964, 2936, 2669, 1731, 1708, 1469, 1382, 1110 cm<sup>-1</sup>; <sup>1</sup>H and <sup>13</sup>C NMR data, see Table 1; EIMS *m/z* 438 [M]<sup>+</sup>, 423, 405; HREIMS *m/z* 438.3485 (calcd for C<sub>30</sub>H<sub>46</sub>O<sub>2</sub>, 438.3498).

**3 $\beta$ -Acetoxyfern-9(11)-en-19-one (2):** white, amorphous powder; [ $\alpha$ ]<sub>D</sub><sup>20</sup> +6.7 (c 0.15, CHCl<sub>3</sub>); IR (KBr)  $\nu_{\max}$  2948, 2854, 1732, 1451, 1376, 1246, 1027 cm<sup>-1</sup>; <sup>1</sup>H and <sup>13</sup>C NMR data, see Table 1; EIMS *m/z* 482 [M]<sup>+</sup>, 467, 407; HREIMS *m/z* 482.3782 (calcd for C<sub>32</sub>H<sub>50</sub>O<sub>3</sub>, 482.3760).

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**Supporting Information Available:** Spectroscopic data of new compounds **1** and **2**. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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